

Using Computer Modeling to Predict and Optimize Separations for Comprehensive 2-Dimensional Gas Chromatography

Frank L. Dorman, Paul D. Schettler, and Leslie A.
Vogt

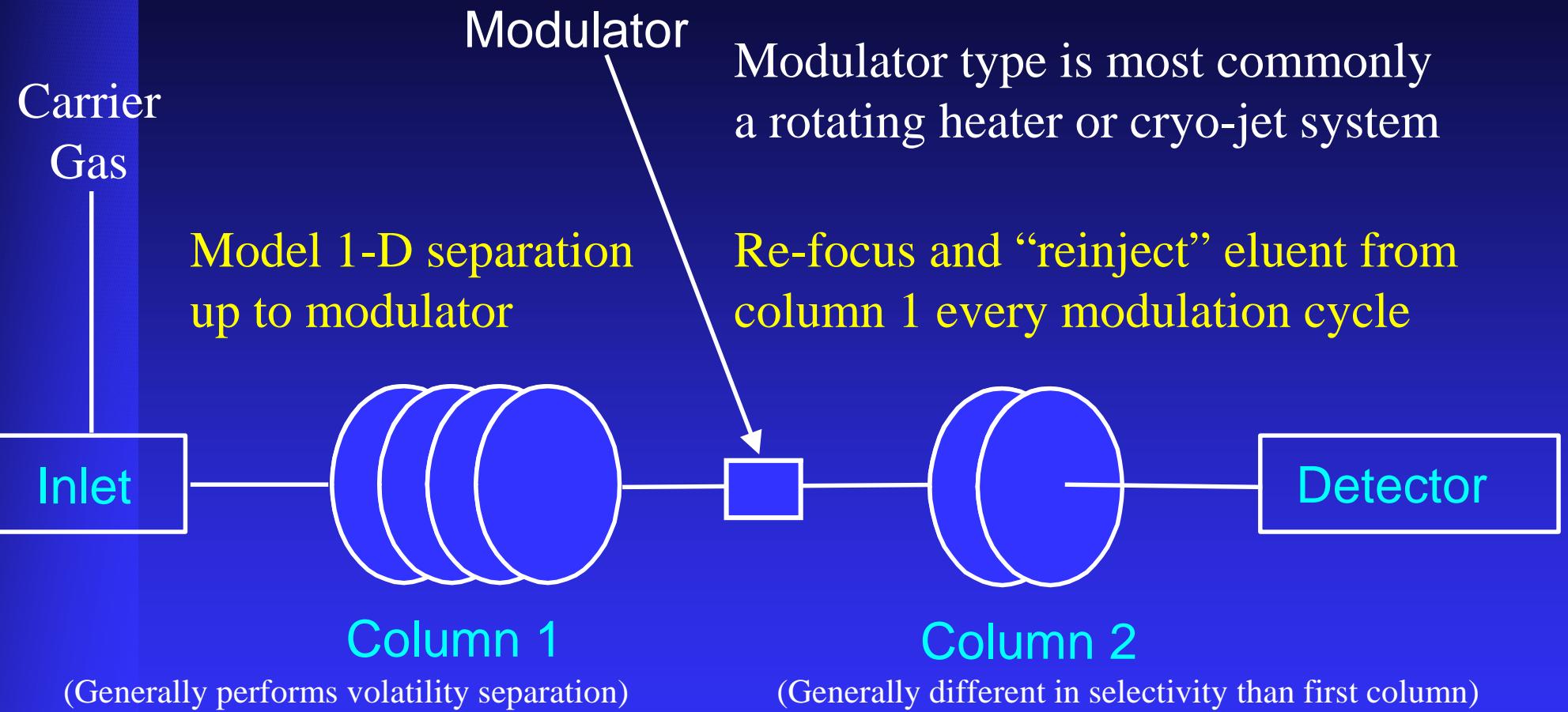
Department of Chemistry, Juniata College,
Huntingdon, PA

Jack W. Cochran
LECO Corporation, Las Vegas, NV

Restek Corporation
www.restekcorp.com



Comprehensive 2D-GC System



Why do we need to model GCxGC Separations?

- Optimization is much more difficult
 - Changes in oven programming, modulation time, offset temperatures, etc... have a more complex impact on separations
- “Translation” from GC-GC-TOFMS for method development to GCxGC-FID or GCxGC-ECD for more routine analysis
- Choice of best column ensemble is not as obvious

Optimization/Modeling of Separations

- Several approaches have been used for conventional separations
- Allows prediction of optimal conditions for a users column (Pro EZ-GC)
- Allows prediction of optimal stationary phase chemistry and conditions (*Anal. Chem.* **74**(9), 2133-2138 2002)
- Can these be applied to Comprehensive 2-D separations?

1-D Modeling

General Equation for Resolution:

$$R = 1/4 \sqrt{L/h} \times (k/k+1) \times (\alpha-1/\alpha)$$

Selectivity Factor (α) – addressed by stationary phase modeling

not commonly done by end user

Capacity Factor (k), and Column Factor – addressed by physical modeling

can be simultaneous with, or independent of stationary phase modeling

Stationary Phase Optimization

- Window diagramming
- Computer simulation of phase selectivity, independent of column dimensions (ezGC™)
- Rtx®-CLPesticides, Rtx-CLPesticides2
- Computer prediction of optimized stationary phase composition AND column dimensions
 - Rtx-TNT Rtx-TNT2, Rtx-VMS, Rtx-VGC, Rtx-5SilMS, Rtx-VRX, Rtx-OPPesticides2, Customer-specific columns
- Computer prediction of solute/stationary phase interactions for new polymer designs

Achieving Analyte Separation

Resolution

$$R = 1/4 \sqrt{L/h} \times (k/k+1) \times (\alpha-1/\alpha)$$

Capacity Factor

$$k = (t_R - t_0) / t_0$$

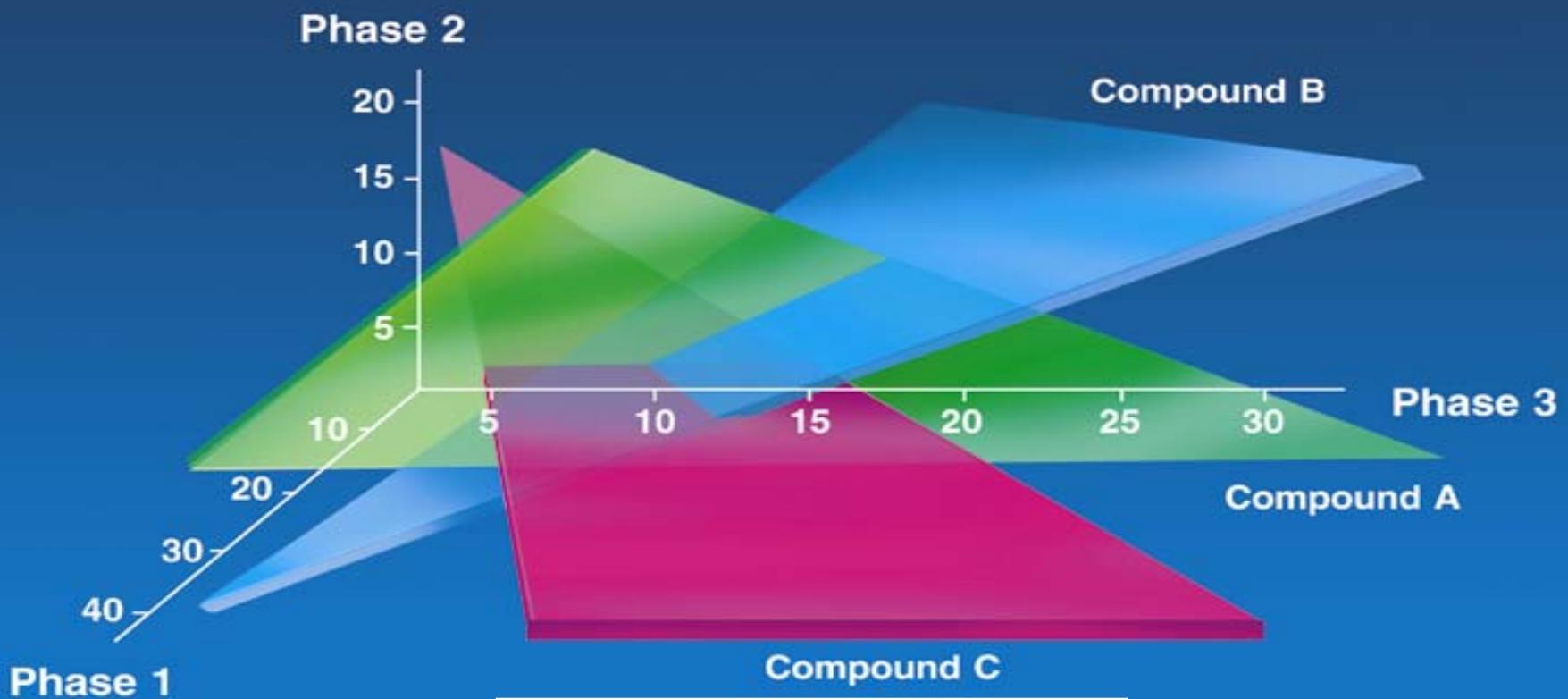
Selectivity

$$\alpha = k_2 / k_1$$

Thermodynamics:

$$\Delta G = \Delta H - T\Delta S \quad \Delta G = -RT \ln K_D$$

3-Space Selectivity Model for 3 Compounds

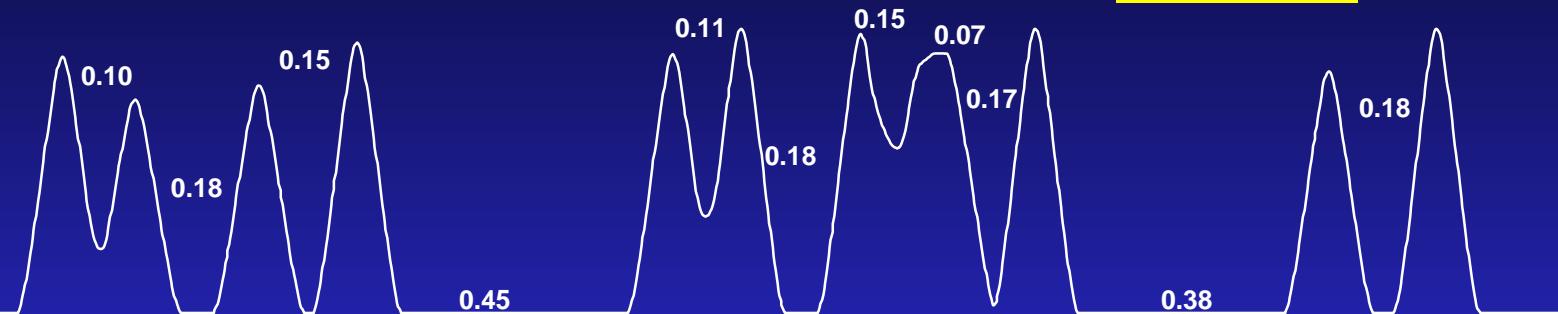


$$\text{Surface} = \mathcal{F}_{\Delta H \Delta S}$$

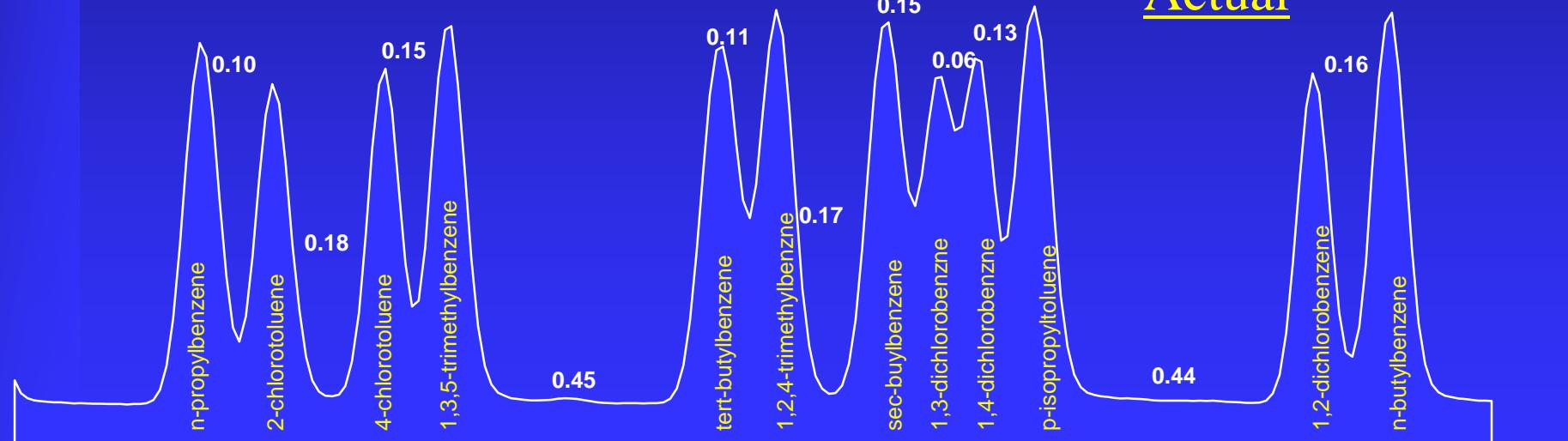
Volatiles Analysis: Predicted vs. Actual 4 Dimensional Phase

Anal. Chem. **74**(9), 2133-2138 2002

Predicted

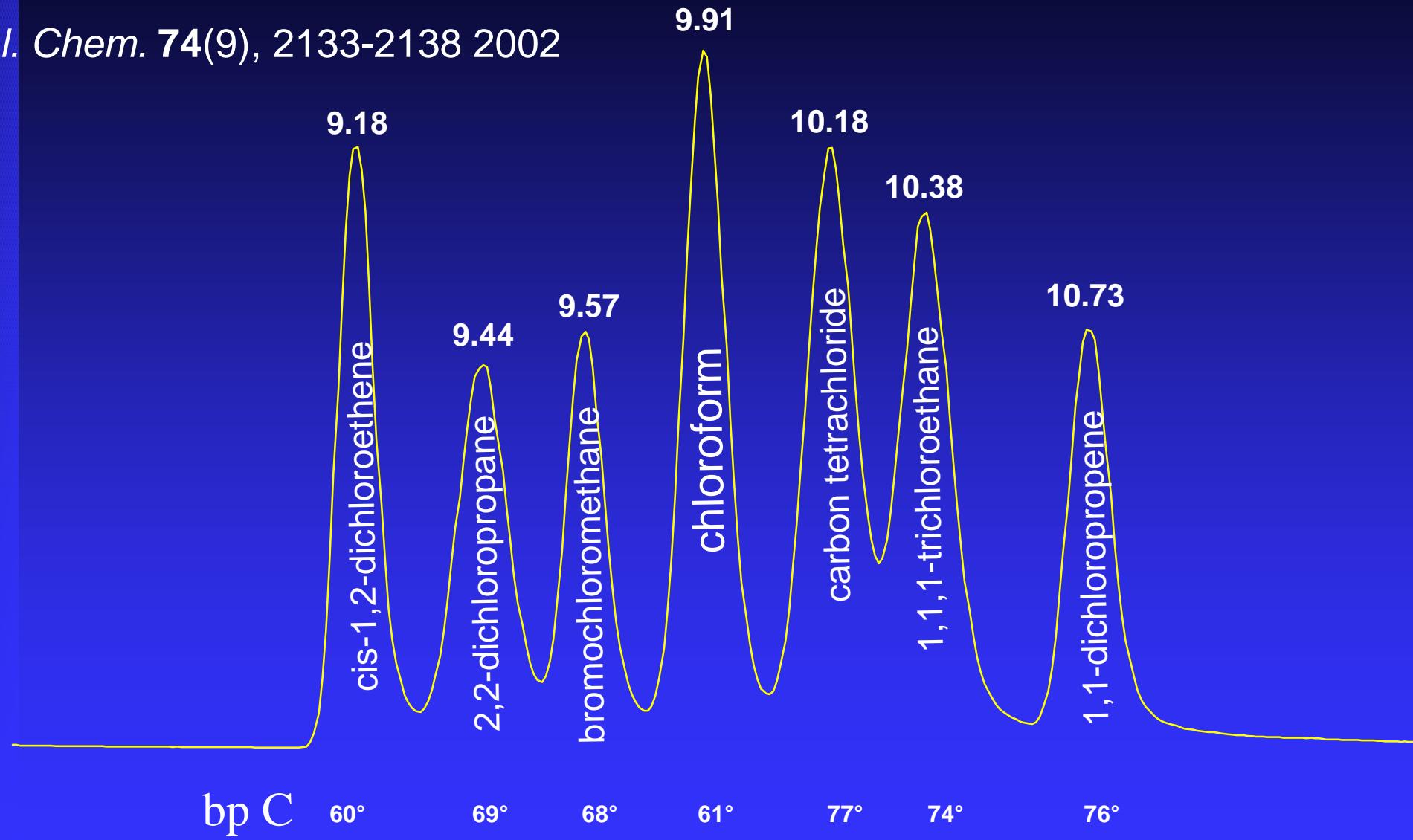


Actual



Volatiles Analysis: Rtx[®]-VGC

Anal. Chem. 74(9), 2133-2138 2002



Simultaneous 2-D Modeling

---or---

Why 2D is not 2X 1D GC

- 1.) Calculation of the pressure at the modulation point
 - not as straight forward during a “desorption”
- 2.) Not a normal “injection” onto second column
 - there may be a selective retention of analyte in the phase at the modulation point
- 3.) There is no “buffer volume” at the modulation point, or a pressure controller to adjust for any pressure surges
 - at 1 atm a liquid expands 1000X as it vaporizes

Input Data

- Compounds of interest analyzed in 1-D mode on each stationary phase of interest at two different temperatures or temperature programs
- A separate ΔH and ΔS are calculated for each compound on each stationary phase independent of physical parameters
- Separation is modeled on conditions of 1st column
- Eluent is re-focused (peak width is re-calculated) and injected onto 2nd column
- Final elution from second column is reported as a function of both 1st and 2nd dimension – tabular report

Grob Test Mixture

As numbered on 2D Chromatogram

- 1.) 2,3-butanediol
- 2.) decane
- 3.) undecane
- 4.) 1-octanol
- 5.) 1-nonanol
- 6.) 2-ethylhexanoic acid
- 7.) 2,6-dimethylphenol
- 8.) 2,6-dimethylaniline
- 9.) C10 FAME
- 10.) C11 FAME
- 11.) dicyclohexylamine
- 12.) C12 FAME

Pressure at Modulation Point

$$P_m^2 = \frac{\pi}{16R} \left(\frac{p_{icoll1}^2 a_2 + p_{ocoll2}^2 a_1}{a_1 + a_2} \right)$$

$$a_1 = \frac{\eta_1 T_1 L_1}{r_1^4} \text{ and } a_2 = \frac{\eta_2 T_2 L_2}{r_2^4}$$

Partition Coefficient of Component i in Stationary Phase i

$$K_i = e^{-(\Delta H_i - T\Delta S_i)/RT}$$

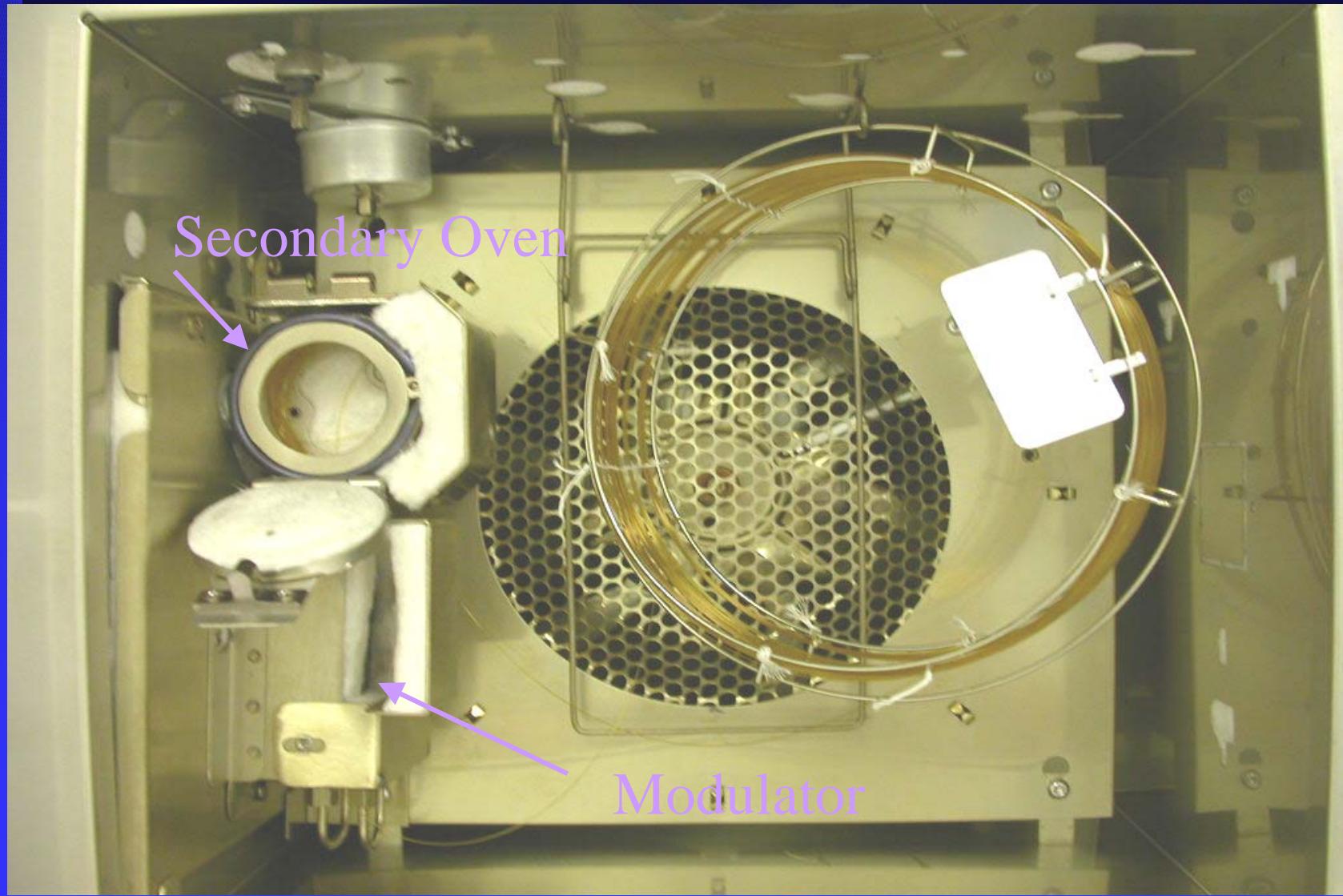
Retention ratio as a function of stationary phase composition

$$\mathfrak{R} = \frac{1}{1 + \beta \sum_{i=1}^N X_i K_i}$$

LECO GC_xGC-FID



Modulator and Secondary Oven



Run Conditions

Primary Oven

Initial Temp (*C)	40
Hold Time (min)	0.2
Ramp (*C/min)	17
Final Temp (*C)	200
Hold Time (min)	1

Modulator

Temp Offset (*C)	30
Mod. Time (sec)	3
Hot Pulse (sec)	0.4

2nd Oven

Temp Offset (*C)	10
------------------	----

GC Conditions

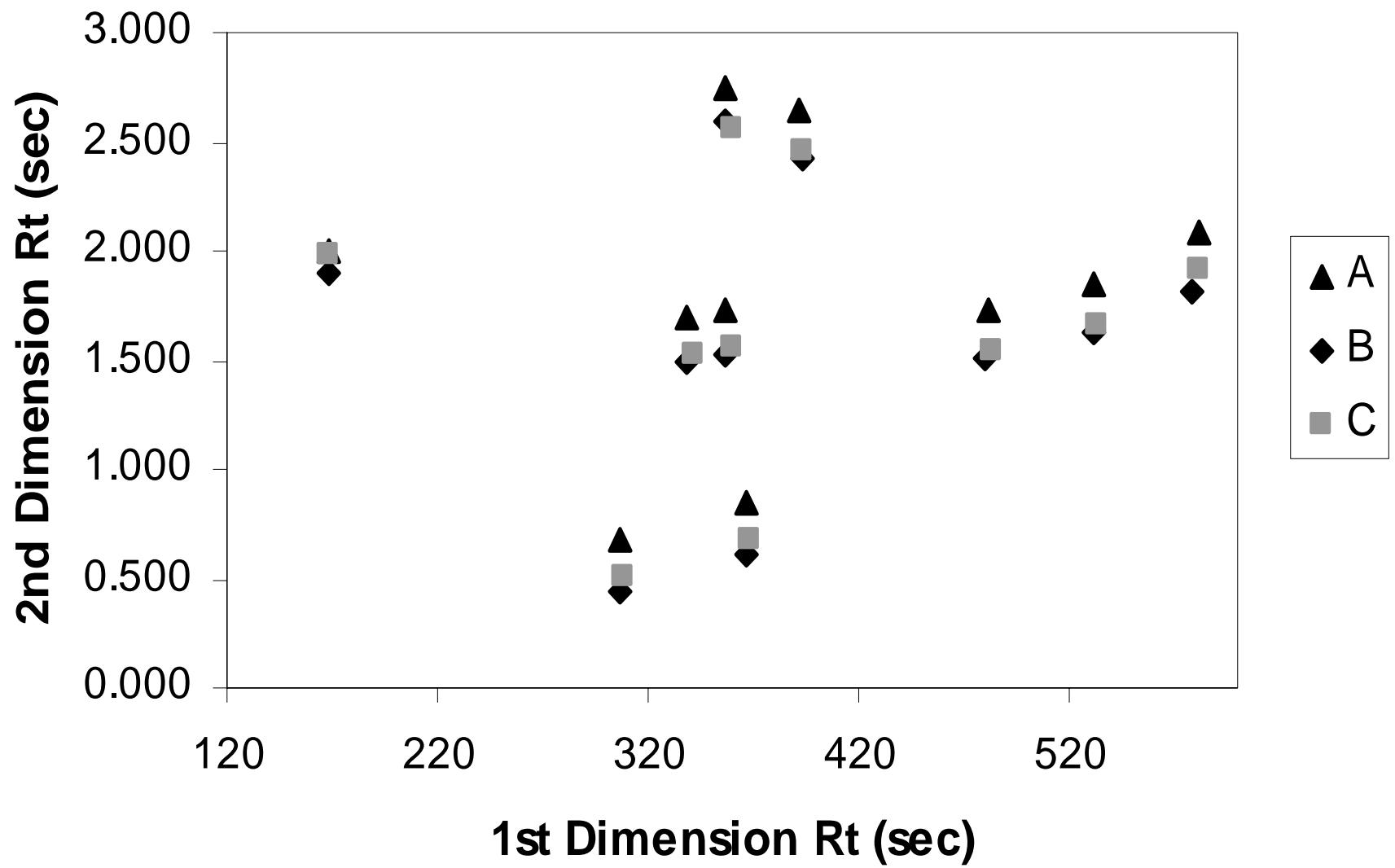
Constant Pressure	20 psi
Split Ratio	10:1
Inlet Temp	250°C
Detector Temp	250°C

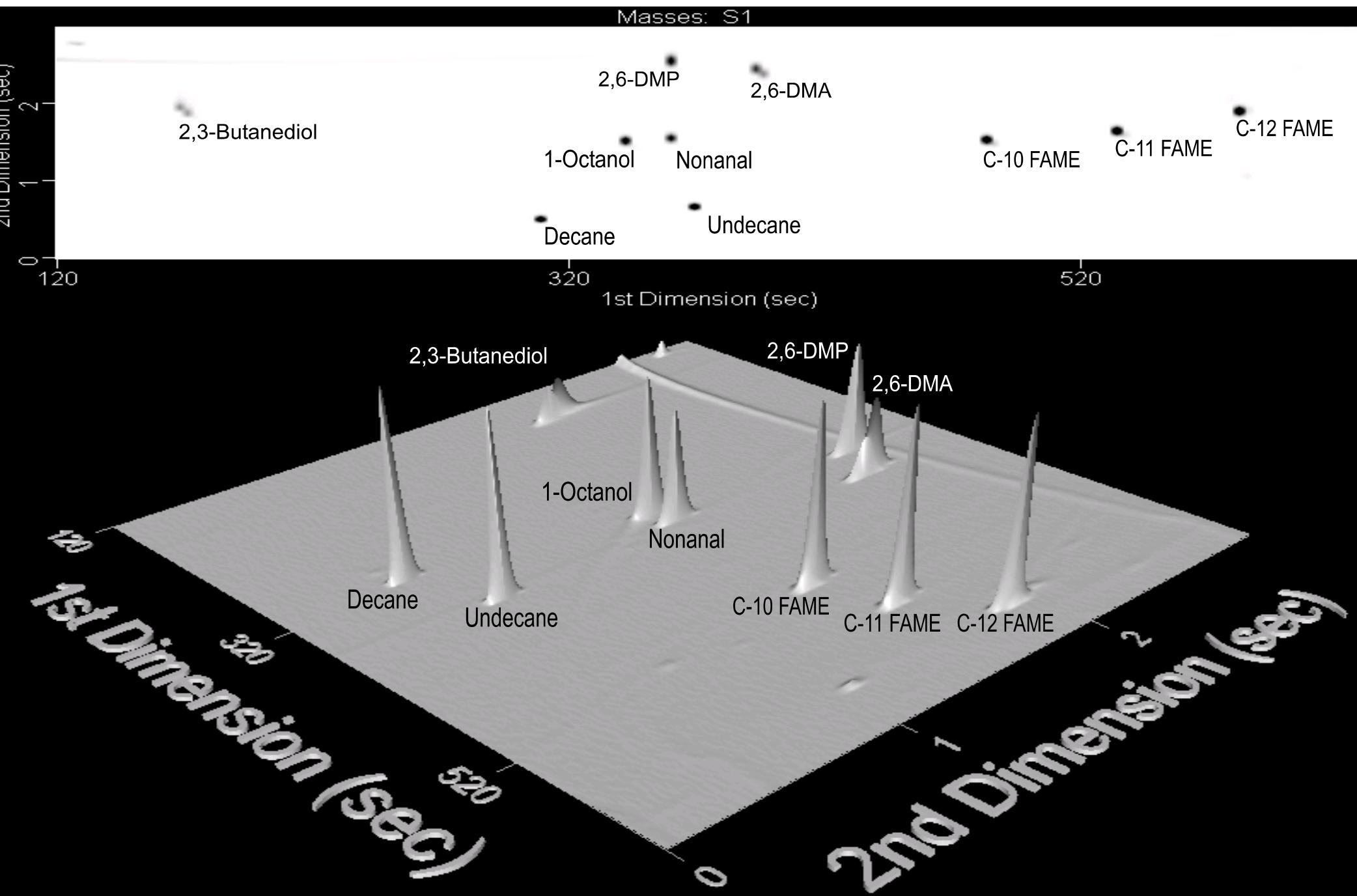
Effect of Modulation Location

- Selective desorption out of stationary phase during modulation cycle
- Tested three configurations:
 - A - Modulation on end of first column
 - B - Modulation on start of second column
 - C - Modulation on intermediate transfer line

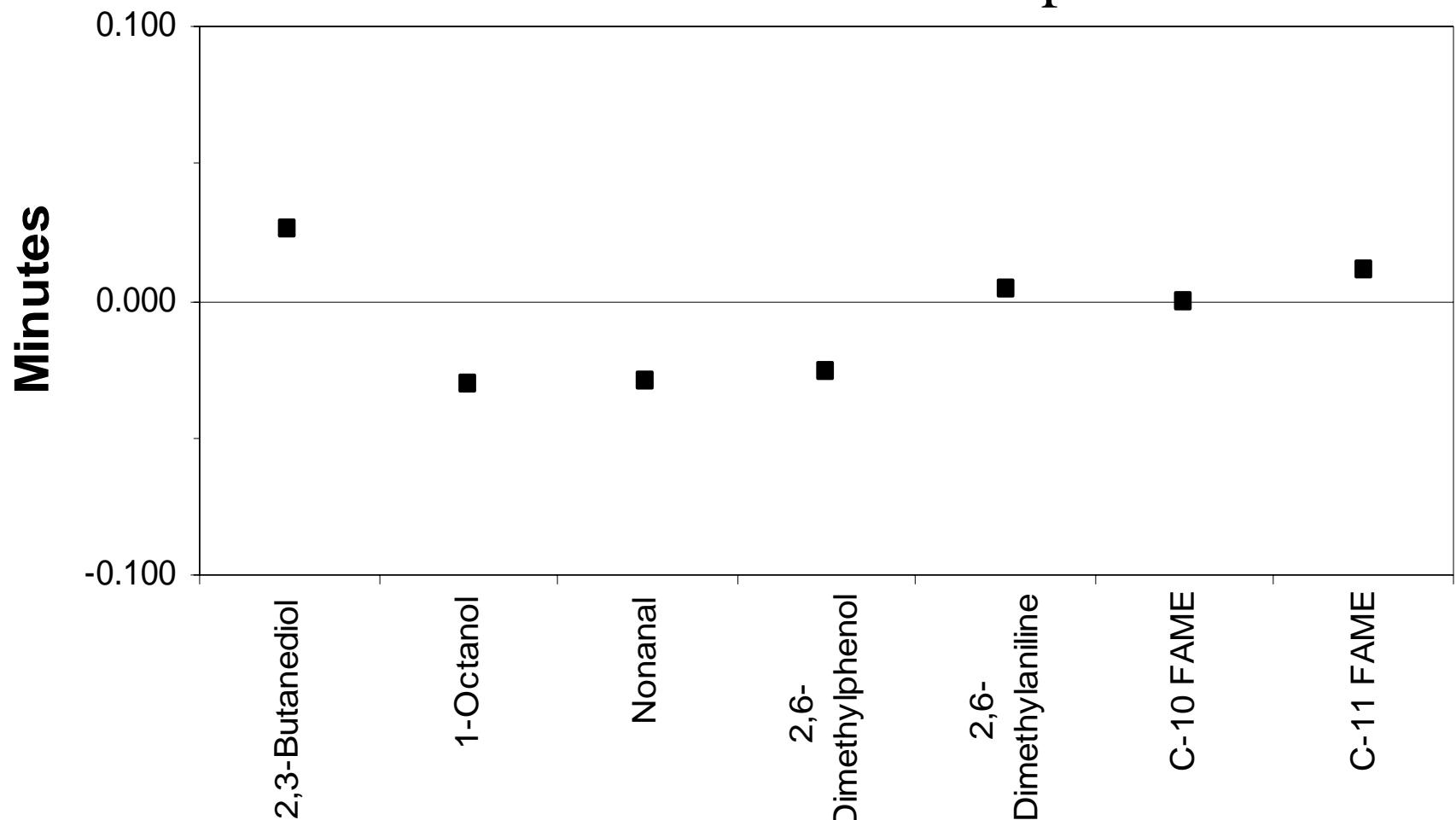
1st Dim Rt	End of 1st	Beg of 2nd	Guard Col
	A	B	C
2,3-Butanediol	168	168	168
Decane	306	306	309
1-Octanol	339	339	342
Nonanal	357	357	360
2,6-Dimethylphenol	357	357	360
Undecane	366	366	369
2,6-Dimethylaniline	391	393	393
C-10 FAME	481	480	483
C-11 FAME	532	531	534
C-12 FAME	581	579	582

2nd Dim Rt	End of 1st	Beg of 2nd	Guard Col
	A	B	C
2,3-Butanediol	1.995	1.905	1.980
Decane	0.678	0.445	0.515
1-Octanol	1.688	1.500	1.533
Nonanal	1.733	1.530	1.568
2,6-Dimethylphenol	2.743	2.588	2.563
Undecane	0.852	0.610	0.680
2,6-Dimethylaniline	2.642	2.420	2.460
C-10 FAME	1.725	1.508	1.540
C-11 FAME	1.845	1.635	1.658
C-12 FAME	2.083	1.818	1.913

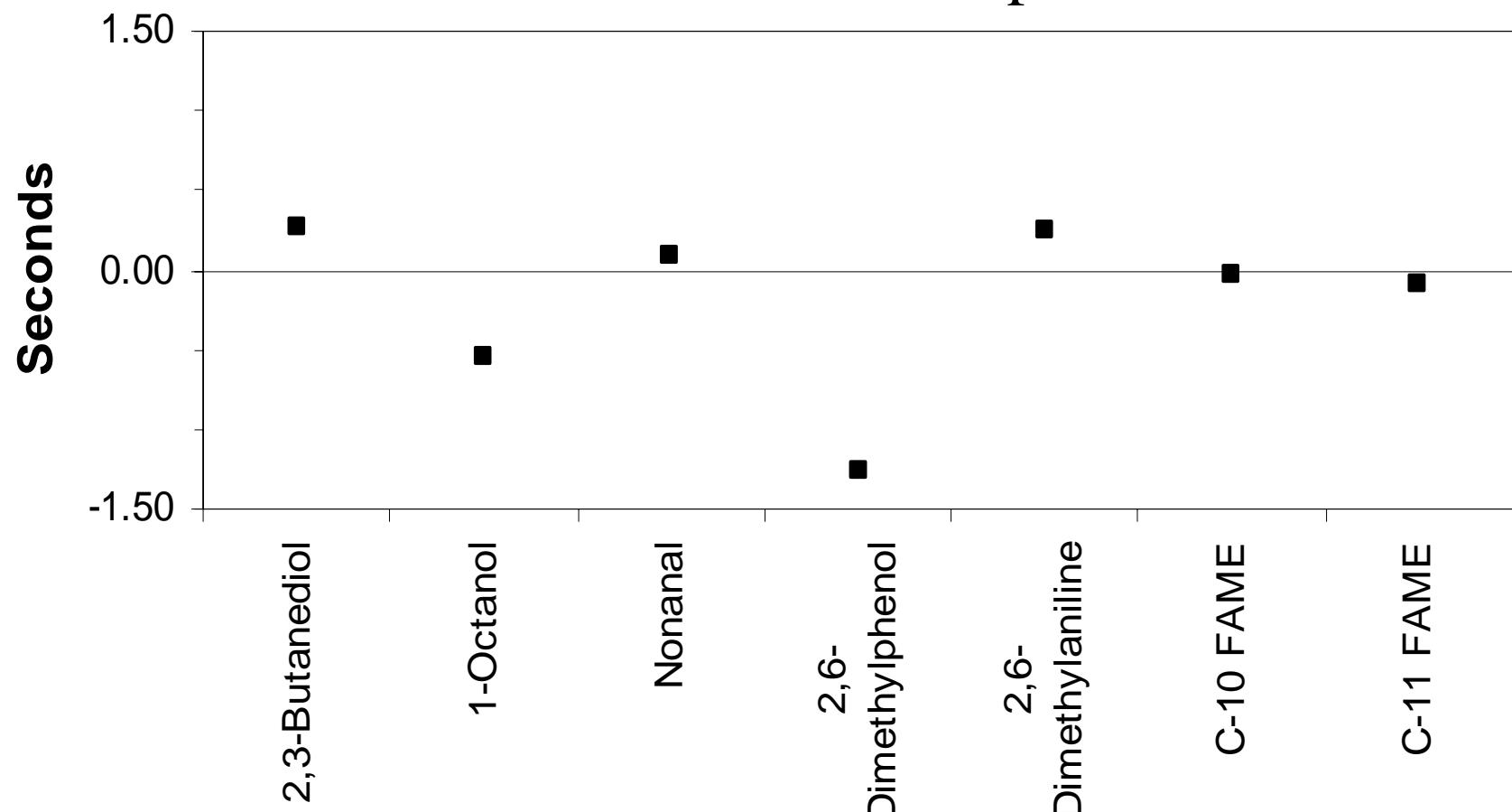




Predicted – Experimental Retention Times, Arrival to Modulation point



Predicted – Experimental Retention Times, Second Dimension Separation



Comparison	(All times in sec)		
	CASPD	CASPD	# Slices
	avg total (sec)	Calc 2nd Rt	(from CASPD)
2,3-Butanediol	5.257	2.2565	1
1-Octanol	3.988	0.9879	1
Nonanal	4.653	1.6531	1
2,6-Dimethylphenol	4.305	1.3047	1
2,6-Dimethylaniline	5.705	2.7054	1
C-10 FAME	4.517	1.5167	1
C-11 FAME	4.578	1.5776	1

	Guard Col	Adj. C	
	C	Total 2nd Rt	CASPD - Adj. C (sec)
2,3-Butanediol	1.980	4.980	0.277
1-Octanol	1.533	4.533	-0.545
Nonanal	1.568	4.568	0.086
2,6-Dimethylphenol	2.563	5.563	-1.258
2,6-Dimethylaniline	2.460	5.460	0.245
C-10 FAME	1.540	4.540	-0.023
C-11 FAME	1.658	4.658	-0.080

Summary

- Modeling procedure appears successful
 - Some refinement is necessary
- Modeling optimization software can save large amount of R&D time for column ensemble development
- Translation between TOFMS and FID/ECD work ongoing
- Simplification of CASPD2D procedures ongoing
- Work submitted to Journal of Chromatography A

Acknowledgements

- LECO Corporation
 - Modulator, Software, Financial Support
- Agilent Technologies
 - 6890 GC loan
- Chemistry Department, Juniata College
 - Financial Support