

Environmental Applications

Analysis of Explosives

by Liquid Chromatography

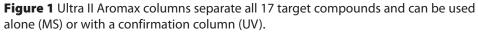
Reliable analysis of explosives is largely dependent on the selectivity and sensitivity of the analytical column. C18 columns are commonly used as primary columns; however, reliable results are difficult to obtain for several critical compounds. Alternative phases were evaluated and the Ultra II Aromax and Ultra C8 columns were determined to provide better separations for the routine analysis of explosives.

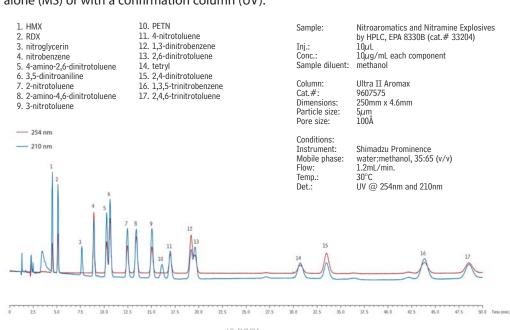
Introduction

Recently, there has been increased interest in highly sensitive explosives analyses for forensic applications, due to the increased emphasis on homeland security. Explosives methods have long been employed by environmental labs focused on soil and groundwater quality in areas where munitions are disposed of through combustion. Target compounds include nitroaromatics, nitramines, and nitrate esters, which present health concerns due to their carcinogenic, mutagenic, and toxic effects. Testing can be performed by gas chromatography (GC), liquid chromatography (LC), or other methods, but LC has several advantages for routine analysis including, ease-of-use, sensitivity, and reliability for thermally unstable compounds.

Analysis of explosives by LC often follows EPA Method 8330B, an update of the original method, which now incorporates mass spectrometry (MS) as an alternative to ultraviolet (UV) detection. An advantage of MS is that it requires only one column, so it is faster as no confirmation analysis is required. MS is also more sensitive than UV; however, when evaluating columns for LC/MS, it is important that the column resolve all critical pairs that may have isobaric interferences. In contrast, UV is less expensive and generally is more readily available in most labs, so both detection techniques are commonly used.

Typically, when analyzing by UV, a C18 primary column with a cyano- or phenyl-based confirmation column is used. However, several compounds can be problematic on these phases, including 2,6-dinitro-toluene and 2,4-dinitrotoluene which may coelute or show poor response with UV detection. Gradients can be used to achieve separation, but this is more time-consuming than using an isocratic method. Tetryl is another difficult explosive compound. It is susceptible to heat degradation and false positives can result from matrix interference causing a retention time shift for 3,5-dinitroaniline (which elutes near tetryl on a C18 column). By using columns with different selectivities, analysts can more accurately identify the compounds of interest. Here we evaluated several alternative LC column phases to determine which phase produced optimal separations for commonly analyzed explosive compounds.





Procedure

Several columns were evaluated for selectivity and retention of the target explosives compounds listed in EPA Method 8330B. Each column was first evaluated for application as a primary UV or stand-alone MS column; secondarily, remaining columns were evaluated for orthogonal characteristics for use as a confirmation column. Orthogonal characteristics included elution order changes and retention time shifts.

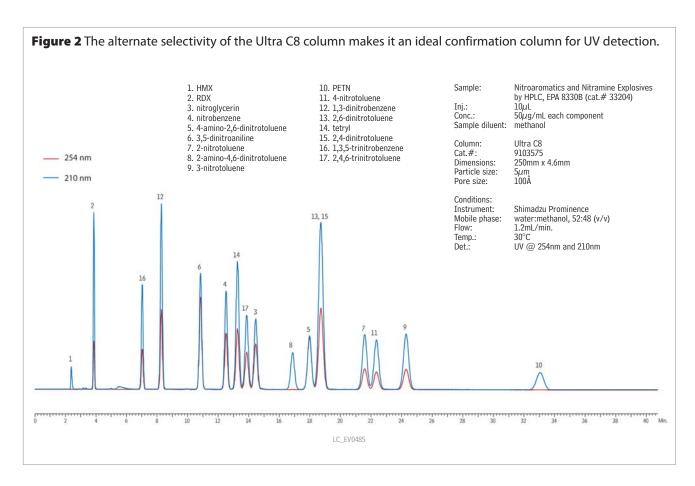
Mixed standards containing all 17 target explosives were prepared in methanol. A simple isocratic elution using methanol and water was used in order to follow the method closely and to eliminate the need to re-equilibrate the columns prior to each analysis (as required when using a gradient elution). Column dimension were consistent with method recommendations of 250mm x 4.6mm, 5µm. Column temperature variations were eliminated by using a column heater set at 30°C. All peak identities were confirmed by comparison to individual standards for each compound.

Results

The Ultra II Aromax column provided the best separation of all columns tested (Figure 1). The unique selectivity of this column allows for the separation of all 17 analytes with no coelutions. All compounds that are typically problematic were well-resolved and positively identified. Since all compounds were resolved on the Ultra II Aromax column, it is recommended for both UV and MS methodology. Further, because the mobile phase has a higher organic content (65% methanol) than is typically used, it has the added benefit of improving MS sensitivity by allowing better desolvation at the mass spectrometer interface. For labs running UV methods, the Ultra C8 column provided several elution order changes and retention time shifts relative to the Ultra II Aromax column, making it an ideal confirmation column for UV analyses (Figure 2). Only a single coelution is seen on the Ultra C8 column, which is fully separated on the Ultra II Aromax column.

Conclusion

Reliable analysis of explosives is largely dependent on the selectivity and sensitivity of the analytical column. The Ultra II Aromax column separates all 17 target analytes in EPA Method 8330B and is compatible with a high organic content mobile phase, which improves performance with MS detectors. If analysis by UV is desired, the Ultra C8 column provides several elution order changes and retention time shifts, making it an excellent confirmation column. The selectivities of the Ultra II Aromax and Ultra C8 columns allow definitive analyte identifications, even for difficult compounds, making them excellent choices for the routine analysis of explosives in both environmental and forensic applications.



Recommended HPLC Columns

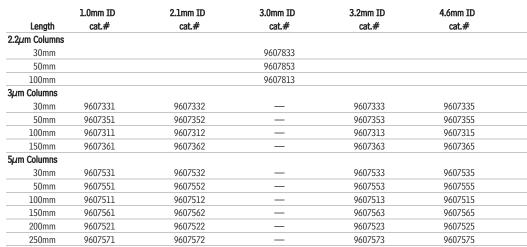
Ultra II Aromax Columns (USP L11)

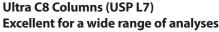
Physical Characteristics:

particle size: 3µm or 5µm, spherical endcap: fully endcapped pore size: 100Å pH range: 2.5 to 7.5 carbon load: 17% temperature limit: 80°C

Chromatographic Properties:

Ultra II Aromax is a unique reversed phase material that exhibits superior retention and selectivity for aromatic and/or unsaturated compounds, compared to conventional alkyl and phenyl phases. This column is a great alternative to our Biphenyl phase when increased retention is required. A very suitable choice for analysis of steroids, tetracyclines, drug metabolites, and other compounds that contain some degree of unsaturation.





Physical Characteristics:

particle size: 3µm or 5µm, spherical endcap: fully endcapped pore size: 100Å pH range: 2.5 to 7.5 carbon load: 12% temperature limit: 80°C

Chromatographic Properties:

A retentive, high-purity, base-deactivated reversed phase packing that exhibits excellent peak shape for a wide range of compounds. Less retention for neutral, hydrophobic compounds, compared to the Ultra C18 column.

	1.0mm ID	2.1mm ID	3.2mm ID	4.0mm ID	4.6mm ID
Length	cat.#	cat.#	cat.#	cat.#	cat.#
3µm Columns					
30mm	9103331	9103332	9103333	_	9103335
50mm	9103351	9103352	9103353	_	9103355
100mm	9103311	9103312	9103313	_	9103315
5µm Columns					
30mm	9103531	9103532	9103533	_	9103535
50mm	9103551	9103552	9103553	_	9103555
100mm	9103511	9103512	9103513	9103514	9103515
150mm	9103561	9103562	9103563	9103564	9103565
200mm	9103521	9103522	9103523	_	9103525
250mm	9103571	9103572	9103573	_	9103575

To order a 2.1mm, 3.2mm, or 4.6mm ID column with a Trident Integral Inlet Fitting, add "-700" to the catalog number for the column.

Example: 100mm x 4.6mm ID Ultra C18 column with Trident Integral Inlet Fitting: 9174315-700 Nominal additional charge.

Visit www.restek.com for guard cartridges for these columns.





Recommended Analytical Reference Materials

To ensure the highest quality explosives standards, Restek chemists carefully purify or synthesize all of the compounds listed in EPA Method 8330.

8330 Internal Standard

3,4-dinitrotoluene 1,000µg/mL in methanol, 1mL/ampul cat. # 31452 (ea.)

8330 Surrogate

1,2-dinitrobenzene 1,000 μ g/mL in methanol, 1mL/ampul cat. # 31453 (ea.)

Nitroaromatics and Nitramine Explosives

by HPLC, EPA 8330B* (17 components)

2-amino-4,6-dinitrotoluene
4-amino-2,6-dinitrotoluene
3,5-dinitroaniline
1,3-dinitrobenzene
2,4-dinitrotoluene
2-nitrotoluene
3-nitrotoluene
4-nitrotoluene
2-nitrotoluene
3-nitrotoluene
4-nitrotoluene
2-nitrotoluene
3-nitrotoluene
4-nitrotoluene
4-nitrotoluene
4-nitrotoluene
4-nitrotoluene
4-nitrotoluene
4-nitrotoluene

HMX 1,3,5-trinitrobenzene nitrobenzene 2,4,6-trinitrotoluene nitroglycerin

 $1,000\mu$ g/mL each in acetonitrile, 1mL/ampul cat. # 33204 (ea.)

Nitroaromatics and Nitramine Explosives

by HPLC* (14 components)

1,3-dinitrobenzene 2-nitrotoluene 2-amino-4,6-dinitrotoluene 4-amino-2,6-dinitrotoluene 4-nitrotoluene 2,4-dinitrotoluene RDX tetryl

HMX 1,3,5-trinitrobenzene nitrobenzene 2,4,6-trinitrotoluene

1,000µg/mL each in acetonitrile, 1mL/ampul cat. # 33905 (ea.)

8330 Calibration Mix #1* (7 components)

1,3-dinitrobenzene
2,4-dinitrotoluene
HMX
2,4,6-trinitrotoluene
nitrobenzene
1,3,5-trinitrobenzene
2,4,6-trinitrotoluene

 $1,000\mu g/mL$ each in acetonitrile, 1mL/ampul cat. # 31450 (ea.)

8330 Calibration Mix #2* (7 components)

2-amino-4,6-dinitrotoluene 4-amino-2,6-dinitrotoluene 2,6-dinitrotoluene 2-nitrotoluene

1,000 μ g/mL each in acetonitrile, 1mL/ampul cat. # 31451 (ea.)

8330 Nitroaromatics Kit (1,000µg/mL)

31450: 8330 Calibration Mix #1 31451: 8330 Calibration Mix #2 31452: 8330 Internal Standard Mix 31453: 8330 Surrogate Mix

Contains 1mL each of these mixtures.
______ cat. # 31454 (kit)

Single-Component Explosives Solutions

Volume is 1mL/ampul. Concentration is $\mu g/mL$.

ACN ACN ACN ACN M	1,000 1,000 2,000 1,000 1,000	31670 31671 31890 31661
ACN ACN M	2,000 1,000	31890
ACN M	1,000	
M		31661
	1.000	
ACN		31453
	1,000	31662
ACN	1,000	31663
ACN	1,000	31664
EA	2,000	33901
M	1,000	31452
M	1,000	31601
ACN	1,000	31665
ACN	1,000	31657
M	1,000	31498
M	1,000	31602
ACN	1,000	31659
ACN	1,000	31660
ACN	1,000	31658
M	1,000	31600
M	1,000	31499
M	1,000	31821
ACN	1,000	31666
ACN	1,000	31667
ACN	1,000	31668
ACN	1,000	31669
	ACN ACN ACN ACN BACN ACN ACN ACN ACN ACN ACN ACN ACN ACN	ACN 1,000 ACN 1,000 ACN 1,000 EA 2,000 M 1,000 ACN 1,000 M 1,000 ACN 1,000 M 1,000 M 1,000 M 1,000 M 1,000 ACN 1,000

ACN=acetonitrile M = methanol

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*Meet all DOT requirements. Available only to customers or distributors inside the 48 contiguous United States; items may not be resold for export.





Lit. Cat.# EVAN1176

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